

ACTIONS CONCERTÉES EN Océanologie

RAPPORT D'ACTIVITE

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IV

ANNUAL VARIATION OF Zn, Cd, Pb AND Cu IN SEAWATER
AND PARTICULATE MATTER IN THE NORTH SEA(SOUTHERN BIGHT),

by: G.GILLAIN

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INTRODUCTION.

Surface water samples, collected over 1 year in the North Sea, are analysed for Zn, Cd, Pb and Cu using pulse polarography directly on seawater. (I-I5).

The purpose of these cruises is to achieve a chemical survey of coastal waters of the North Sea.

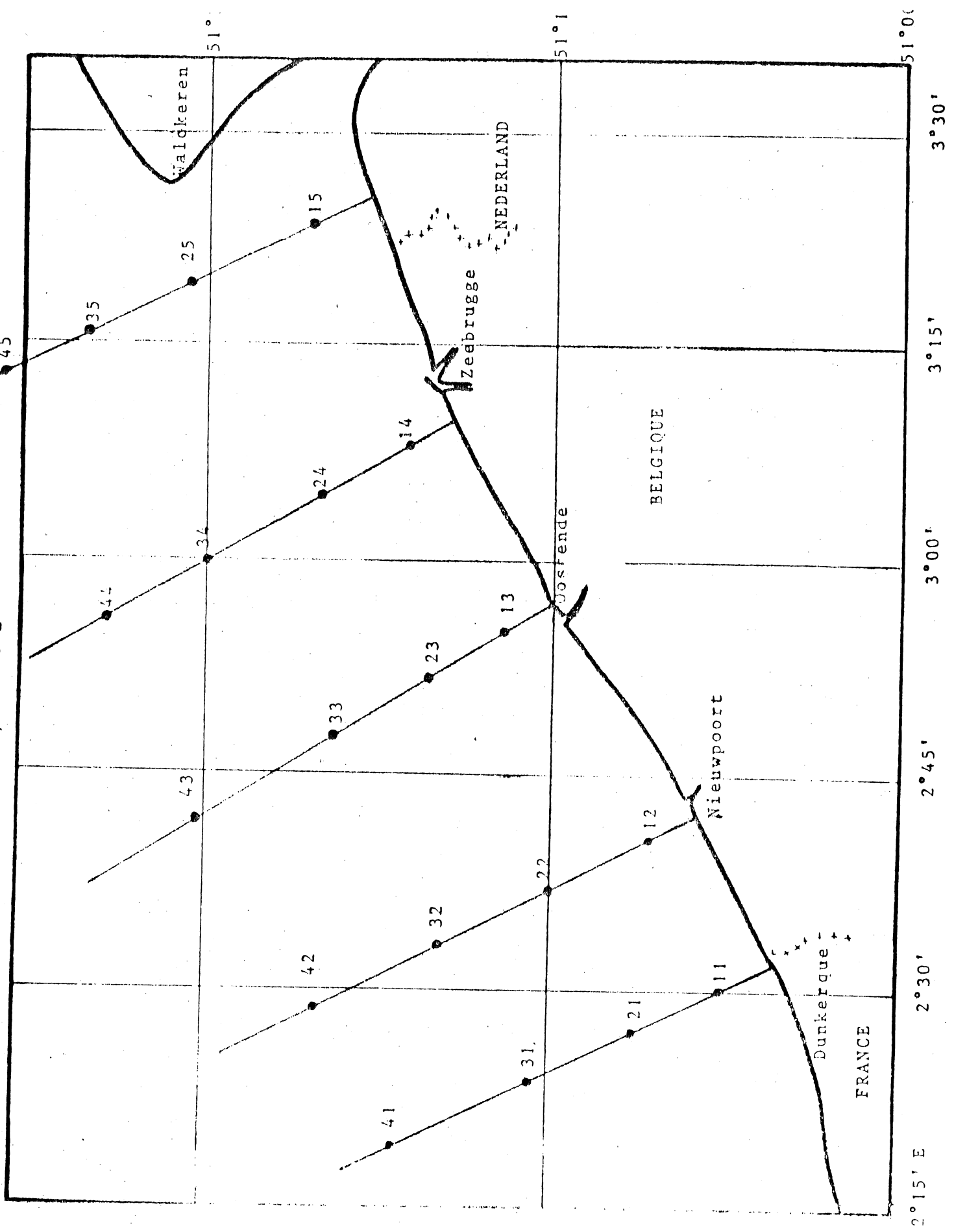
The location of the stations where samples are collected is summarized in table I and reported on map I.

TABLE I

Code of number
of stations.

11	51°07'20" N	02°30'10" E
21	51°11'10" N	02°27'20" E
31	51°15'40" N	02°23'40" E
41	51°20'40" N	02°19'20" E
I2	51°10'30" N	02°40'30" E
22	51°15'00" N	02°37'10" E
32	51°19'40" N	02°33'20" E
42	51°25'00" N	02°29'00" E
I3	51°16'40" N	02°54'50" E
23	51°19'50" N	02°52'00" E
33	51°24'00" N	02°48'00" E
43	51°30'00" N	02°42'00" E
I4	51°21'25" N	03°07'40" E
24	51°25'10" N	03°03'20" E

MAPE I



34	51°30'00"	03°00'00"
44	51°34'20"	02°55'40"
15	51°25'00"	03°24'00"
25	51°31'00"	03°19'20"
35	51°35'20"	03°15'40"
45	51°39'25"	03°12'25"

Station code number: 1st number = number of the station on
the radial,
2^d number = number of the radial.

METHODS.

1) Sampling, filtration and storage (II)

Water samples are collected with a oceanographic reversing bottle at a depth of 3 m and filtered immediately through Millipore filters (pore size: 0.8 μ m) washed in 10^{-2} M DTPA solution and rinsed with tridistilled water before the filtration. One liter of solution is transferred to acid-cleaned high density polyethylene bottles rinsed with seawater which are then stored with the filters (suspended matter) in a deep-freezer (-20°C) until the analysis begins.

2) Analysis.

a) dissolved metals.

Concentrations of dissolved Zn, Cd, Pb and Cu in North Sea water samples have been measured at natural pH and acid pH (pH ~ 3), without and after U.V. irradiations by differential pulse anodic stripping and using a hanging mercury drop electrode (KEMULA type).

For the purpose of clarity, this technique will be summarized briefly:

a 30 ml seawater sample is added to the electrolytic cell and

O₂ is removed with N₂. Zn, Cd, Pb and Cu in the sample are then

deposited by preelectrolysis at -1.200 V for a given period of time(10-15 min.) After deposition, the reduced metals in the amalgam are stripped by oxidation and the resulting currents are measured. The current peaks at approximately -1.00 V(Zn), -0.60 V(Cd), -0.40 V(Pb) and -0.12 V(Cu) vs. saturated KCl calomel electrode(SCE) are proportionnal to the amount of metal plated into the mercury drop and to the metal concentration in solution. Then 50-100 μ l of metals standard solution are added to the sample and allowed to equilibrate with the seawater components for a period of 5 minutes. The electrolysis and stripping process are repeated and the metal concentration is obtained from the ratio of the peak current of the sample to that of sample plus the standard.

b)particulate metals .
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-ashing of the samples (II).
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The filters are dried at 60°C and ashed by Low Temperature Asher with micro-wave activated oxygen(LTA Tracerlab).

-analysis of the residue of ashing.
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The residue from the LTA ashing is dissolved in two ml of Suprapure concentrated HCl and is transferred to a 50 ml volumetric flask which is filled with distilled water. 30 ml of the final solution are transferred to the cell for the anodic stripping voltammetry(see above).

RESULTS.

a)dissolved metals.

During the period of july 77 to december 77 the variation of dissolved ions, weakly complexed cations and strongly complexed cations in the North Sea have been investigated(tables II to VI and fig. I).

The organically bound metal fraction in the dissolved state is very

CRUISE OF JULY 1977

Samples	Zn (ppb)			Cd (ppb)			Pb (ppb)			Cu (ppb)		
	I	2	3	I	2	3	I	2	3	I	2	3
II 270777 I055	5,55	10,00	27,00	0,40	0,00	0,14	2,60	1,23	6,08	2,10	6,40	0,00
4I 270777 I410	8,10	12,50	20,36	0,36	0,00	0,33	2,20	1,60	9,80	2,70	2,10	0,60
3I 270777 I300	8,10	2,87	-	0,18	0,33	3,50	1,04	1,40	17,35	1,11	6,31	6,00
42 270777 I505	4,18	6,00	20,17	0,12	0,00	0,62	1,60	1,68	4,05	2,00	4,48	0,00
44 280777 I550	6,94	1,00	15,10	0,10	0,00	0,10	2,18	1,00	4,62	3,80	6,40	0,00
35 280777 2140	9,44	2,68	-	0,18	0,00	-	4,68	4,00	-	5,82	9,30	-
I3 270777 I930	9,95	7,42	28,22	0,13	0,00	0,10	2,70	4,10	27,00	2,63	2,10	5,23
32 270777 I550	8,44	4,10	-	0,37	-	-	3,10	5,00	-	3,18	3,70	-
I4 280777 I810	8,20	4,40	0,00	0,17	0,34	0,00	1,33	1,60	3,00	5,63	4,60	0,00
2I 270777 I215	9,90	6,90	0,00	0,13	0,00	0,39	1,50	5,60	28,00	3,40	9,40	0,00
I2 270777 I745	-	-	-	0,12	0,00	0,00	2,53	2,00	0,00	1,80	2,00	0,00
45 270777 I830	5,60	10,80	0,00	0,11	0,20	0,13	2,30	2,00	14,30	2,30	3,50	5,85
33 270777 2030	9,62	5,90	11,80	0,17	0,00	0,00	2,80	1,50	4,40	3,10	2,20	3,50
I5 280777 I930	12,60	8,20	13,40	0,10	0,07	0,25	1,20	3,00	7,10	1,50	4,15	9,20
45 280777 2220	1,70	2,60	9,10	0,10	0,20	0,13	2,15	5,23	3,05	2,21	7,52	0,00

I : Ionic species,

2 : Weakly complexed species,

3 : Strongly complexed species,

Samples	Zn(ppb)			Cd(ppb)			Pb(ppb)			Cu (ppb)		
	I	2	3	I	2	3	I	2	3	I	2	3
35 070977 I700	2,74	2,29	10,18	0,12	0,00	0,00	1,10	5,06	1,40	2,63	3,90	4,30
33 050977 I440	1,23	5,42	6,31	0,63	0,22	0,39	3,16	1,50	4,39	6,89	5,59	-
I4 060977 I830	1,27	5,92	16,31	0,20	0,86	0,00	1,62	3,00	3,47	3,94	9,61	-
43 050977 I335	-	-	-	0,16	0,00	0,00	0,57	3,05	0,00	9,33	5,00	0,00
41 080977 0850	5,31	-	-	0,24	0,00	0,00	3,45	2,50	6,86	5,57	3,88	0,00
21 060977 I408	4,68	6,33	10,12	0,06	0,20	0,00	1,70	5,00	3,22	2,96	1,76	8,37
34 070977 I340	7,84	7,20	25,80	2,72	1,00	1,96	1,60	3,60	9,98	9,70	-	-
31 080977 I010	2,56	5,04	17,00	0,30	0,00	0,46	1,31	5,60	9,60	4,00	2,08	0,00
45 070977 I600	1,42	3,85	5,00	0,19	0,30	0,00	2,49	7,50	0,00	3,04	2,10	7,65
44 070977 I430	2,42	7,13	18,59	0,44	0,00	0,15	2,96	1,50	2,10	4,05	9,36	0,00
42 080977 0735	2,66	5,53	17,74	0,14	0,00	0,32	1,32	3,30	9,60	2,14	9,00	8,90
I2 060977 0840	3,94	2,40	7,60	2,60	0,00	2,20	9,12	-	-	4,85	8,10	2,30
I5 010977 I940	9,25	8,95	-	0,20	0,00	0,10	3,03	5,00	0,00	3,96	5,20	0,00
I3 050977 I633	2,00	5,80	3,58	0,23	0,13	0,13	4,40	2,70	0,00	7,00	4,70	0,00
32 060977 I110	2,80	8,70	4,17	0,15	0,65	0,42	2,10	5,00	6,00	2,50	5,70	-
II 060977 I555	3,46	9,20	-	0,26	0,19	-	2,80	3,00	-	5,40	9,10	-

I : ionie species,

2 : weakly complexed species,

3 : strongly complexed species,

Samples	Zn			Cd			Pb			Cu		
	I	2	3	I	2	3	I	2	3	I	2	8
I3 03IO77 I240	46	27	56	0,61	0,00	0,00	2,38	10,85	00,00	5,40	6,60	28,50
33 03IO77 IO30	7	10	-	0,60	I,20	0,50	I,70	2,20	6,70	6,75	10,80	14,75
43 03IO77 I4I5	9	7	10	0,40	0,00	0,00	I,90	I,80	3,60	3,70	2,77	3,50
23 04IO77 I4I5	15	73	36	0,60	0,30	0,20	2,32	6,90	2,40	4,10	2,60	2,50
II 04IO77 IO00	69	42	31	I,65	I,00	I,10	2,55	6,10	12,30	8,15	9,00	4,20
2I 05IO77 I420	42	43	41	0,44	0,00	0,22	2,13	I,70	7,70	I,63	4,00	3,60
3I 05IO77 I245	290	397	1190	-	-	-	I,50	3,10	3,90	6,90	4,30	7,75
4I 05IO77 II20	10	23	40	0,13	0,10	0,15	I,30	I,90	2,90	2,25	2,75	4,10
I2 05IO77 I550	T27	53	32	0,70	0,20	0,10	I,30	2,70	0,00	3,00	3,56	2,45
25 06IO77 I400	27	42	26	0,53	0,14	0,90	0,56	4,13	0,00	5,70	4,90	0,00
35 06IO77 I200	51	21	54	0,70	0,00	0,00	I,52	4,30	6,35	2,70	4,30	6,60
45 06IO77 II00	227	253	100	I,30	0,00	0,00	2,30	I,30	14,80	4,50	2,30	12,10
I4 07IO77 I300	320	230	121	0,96	I,20	0,00	3,50	4,70	-	2,10	4,80	7,20
24 07IO77 II00	210	326	1564	I,50	I,50	0,00	5,16	10,16	20,20	3,14	4,20	2,80
34 07IO77 IO00	281	196	697	0,50	I,00	I,00	4,10	5,00	5,80	2,20	4,60	11,00
44 07IO77 0900	54	97	24	I,40	I,20	I,05	3,00	4,13	3,40	2,30	3,00	2,10
Cap gris nez	6,35	6,80	10,65	0,18	0,20	0,00	I,60	I,70	I,40	4,85	2,80	2,90
I9IO77 II30	-	-	-	0,45	0,20	0,60	2,25	I,77	3,00	3,70	2,30	6,00
Hansweert												
2II077 0930												

I: Ionic species; 2: weakly complexed species; 3: strongly complexed species

CRUISE OF NOVEMBER 1977

Samples	Zn			Cd			Pb			Cu		
	I	2	3	I	2	3	I	2	3	I	2	3
13 23II77 0930	-	-	-	0,31	0,17	0,00	1,78	0,90	1,80	5,20	10,70	0,00
14 23II77 1200	3,50	10,50	47,50	0,24	0,34	0,10	1,20	2,60	1,60	0,56	2,00	3,10
14 23II77 1415	1,80	8,20	9,90	0,10	0,17	1,40	1,33	2,00	3,21	1,43	2,10	1,40
22 24II77 1135	1,55	7,10	20,90	0,14	0,38	0,40	1,62	5,35	6,80	3,50	3,20	2,00

I : ionic species

2 : weakly complexed species,

3 : strongly complexed species,

CRUISE OF DECEMBER 1977

Samples	Zn			Cd			Pb			Cu		
	I	2	3	I	2	3	I	2	3	I	2	3
II.I4I277.I030	9,40	34,00	25,50	0,52	0,34	0,00	I,20	3,55	I,00	3,50	2,00	I,20
2I.I4I277.I2I5	8,50	IO,IO	4I,65	0,22	0,00	0,38	0,94	4,68	I5,00	2,42	2,32	I2,50
I2.I5I277.I200	IO,IO	4,IS	24,00	0,20	0,00	0,16	2,00	5,25	6,40	2,60	3,IO	2,80
22.I5I277.I240	7,00	5,90	60,20	0,IO	0,80	I,00	I,00	3,20	20,80	3,70	5,25	IO,IO
42.I5I277.I355	3,50	3,40	28,50	0,30	0,20	0,50	I,30	I,60	I,80	2,IS	3,20	I,50
43.I2I277.I300	8,40	20,50	44,80	0,75	0,20	0,00	2,IO	I,50	2,00	4,70	3,30	2,30
I5.I3I277.II05	I5,40	25,80	I6,50	I,00	0,00	0,00	0,90	3,40	4,20	6,35	0,00	0,00
44.I6I277.I235	4,86	IO,50	5,70	0,23	0,00	0,00	I,40	3,90	I2,40	7,IO	-	-
45.I3I277.I330	5,20	2,20	I2,50	0,27	0,18	0,00	3,IS	I,55	2,00	6,50	3,IO	I5,20
24.I6I277.II20	4,20	I5,36	IO,60	0,32	0,50	0,60	3,50	2,60	3,IO	7,40	4,IO	3,IO
I3.I2I277.I345	33,40	33,30	0,00	0,50	0,00	0,00	2,IO	3,00	4,00	4,70	0,00	0,00
I4.I6I277.I033	5,40	3,50	II,00	0,46	0,00	0,40	I,00	3,70	6,50	5,00	2,50	0,00
35.I3I277.I430	9,20	20,90	0,00	0,50	0,40	0,90	2,IS	2,00	4,00	5,IO	2,80	3,IS
25.I3I277.I225	6,80	5,IO	4,IO	0,35	0,00	0,40	I,23	3,00	20,00	2,70	I,70	2,00
33.I2I277.I350	8,70	5,70	22,00	0,II	0,20	2,00	I,00	2,00	I3,20	0,80	I,20	0,80

I : ionic species,

2 : weakly complexed species,

3 : strongly complexed species,

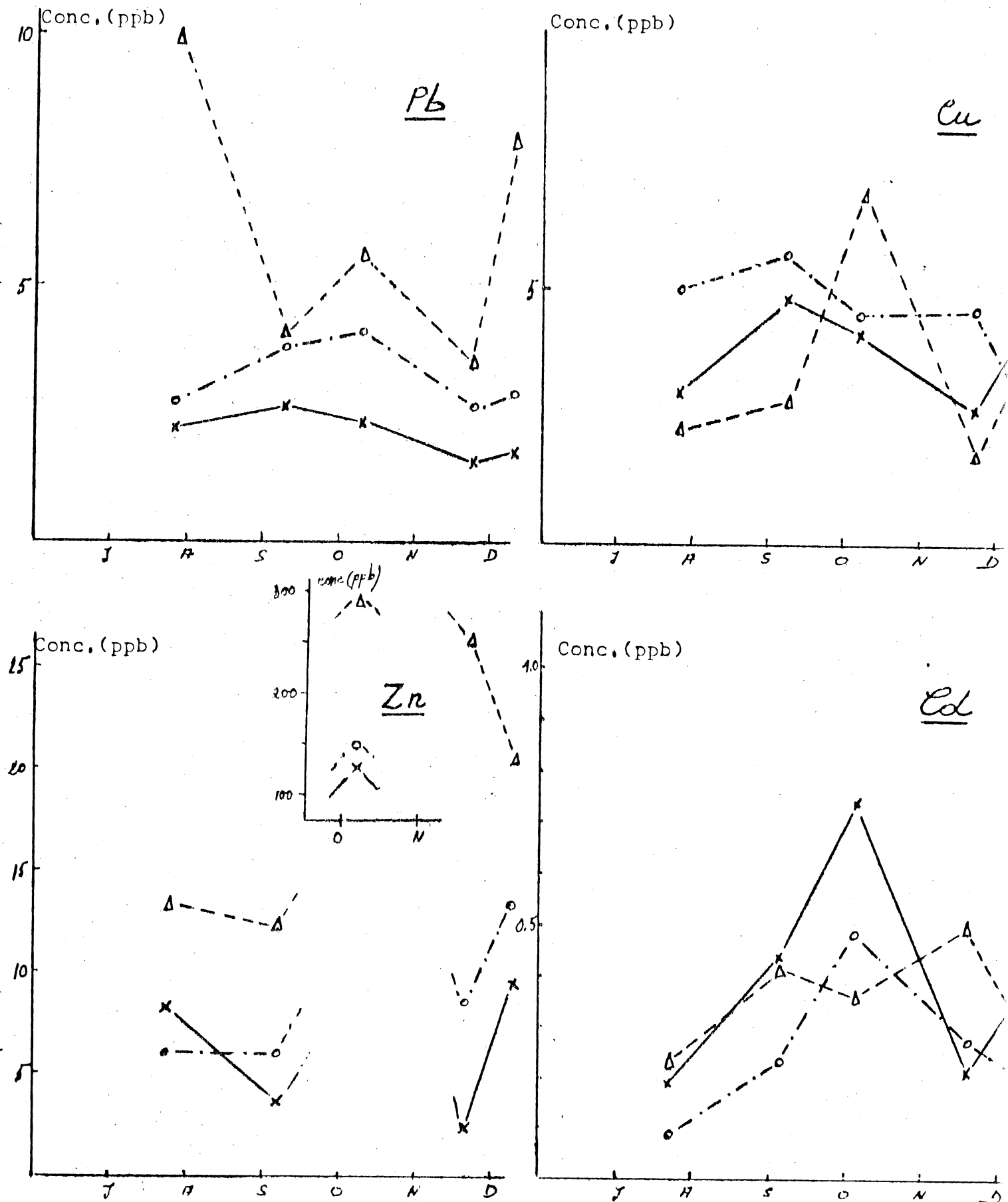


Fig. 1: The concentrations of Zn, Cd, Pb and Cu in surface water of North Sea collected over 1 year period.

- x - ionic species; - o - weakly complexed species; - Δ - strongly complexed species.

Suspended Matter

concentration ppm(dry weight)

Samples	Zn(ppm)	Cd(ppm)	Pb(ppm)	Cu(ppm)
25.300377.I330	I99	2,9	49	64
II.3I0377.II45	I40	0,9	26	-
3I.3I0377.I540	-	0,06	52	94
2I.3I0377.I240	3500	IO	800	4I00
I4.260477.II00	I	-	I	36
25.280477.II50	I74	0,9	26	-
35.280477.I430	225	0,5	44	-
I4.230577.I030	336	0,2	60	87
24.230577.I200	5I5	I,6	7I	I43
34.230577.I400	I33	0,5	29	I47
44.230577.I3I5	-	-	-	-
I2.240577.I030	277	0,I0	86	393
22.240577.II30	I64	0,4	8	23
II.260577.II35	87	3,2	46	II4
2I.260577.I230	3496	3,7	540	-
3I.260577.I430	-	I4,7	94	I25
4I.260577.I345	8I	0,6	II	9I
34.280577.I400	225	0,5	43	-
II.270777.I055	I6	8,8	-	2I4
2I.270777.I2I5	I07	I7	II4	297
3I.270777.I300	40	0,7	II0	I66

concentration ppm(dry weight)

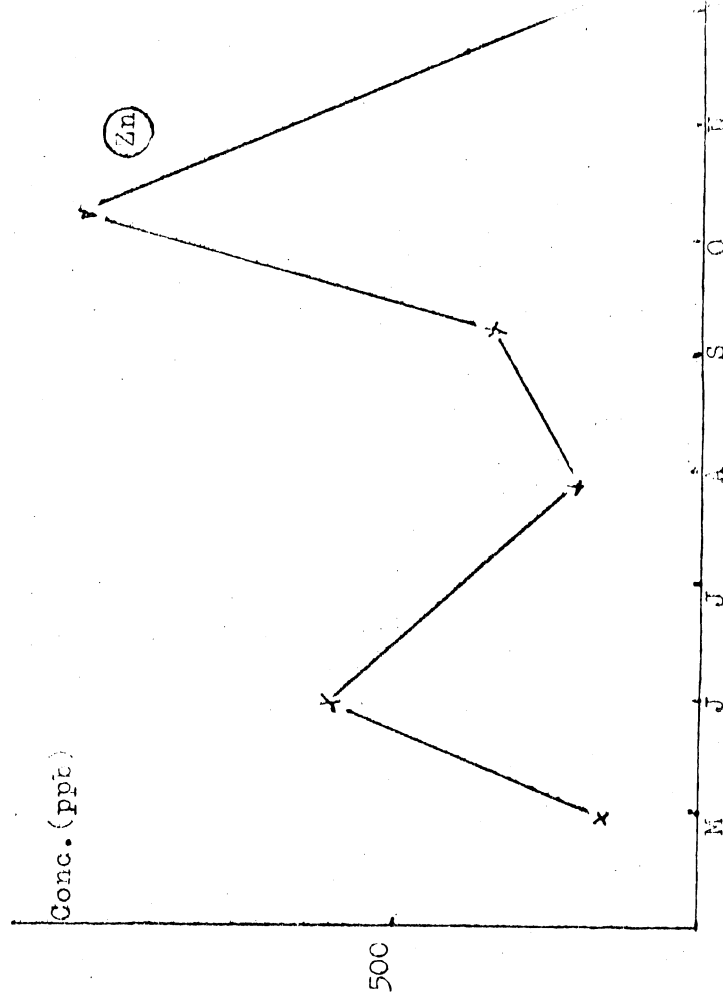
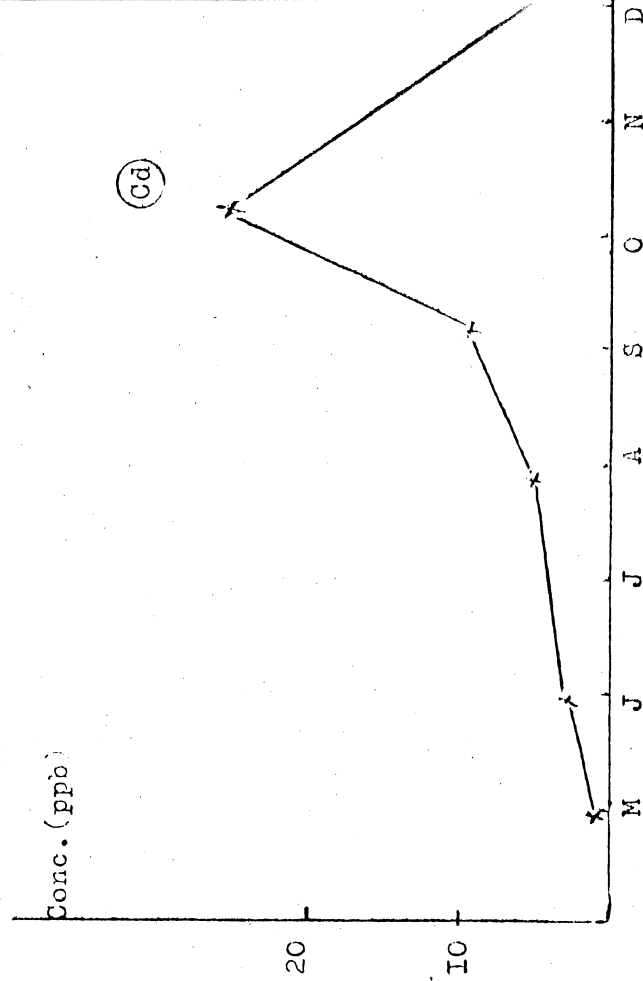
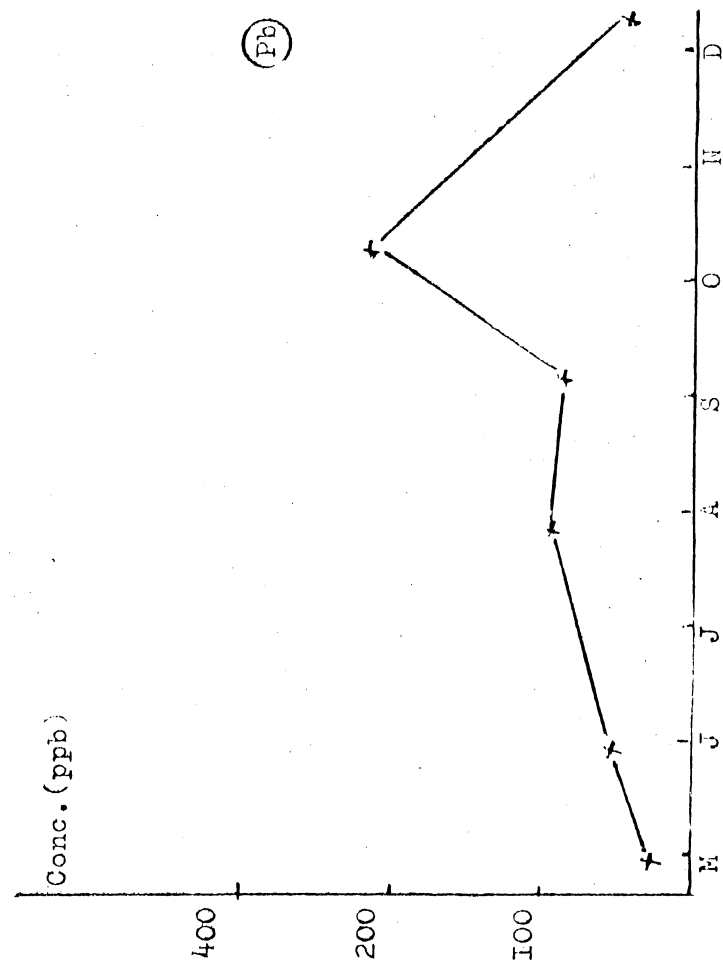
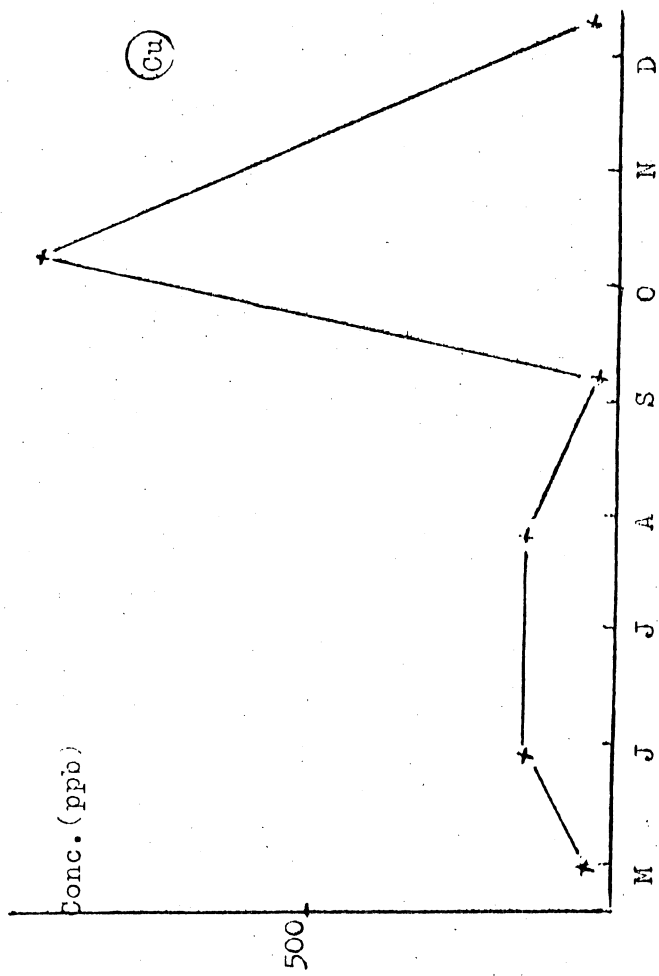
Samples	Zn(ppm)	Cd(ppm)	Pb(ppm)	Cu(ppm)
41.270777.I410	334	7,3	46	103
12.270777.I740	25	0,3	17	30
42.270777.I455	81	0,5	32	188
13270777.I915	I	0,4	172	16
33.270777.2030	235	2,5	46	150
43.270777.I810	166	1,5	39	-
14.280777.I810	166	1,5	39	-
24.280777.I725	142	0,1	6	9
44.280777.I550	153	0,6	58	98
35.280777.2140	-	-	-	-
45.280777.2220	1190*	2,1	253*	194
13.050977.I630	174	0,8	93	332
33.050977.I340	150	1,2	76	250
43.050977.I335	I	0,7	108	70
114060977.I550	-	-	-	-
12060977.0840	166	-	54	174
14.060977.I830	536	5	176	157
15.070977.I940	1230*	3,5	45	98
24.070977.I230	250	6	80	264
34.070977.I340	268	61*	54	210
35.070977.I700	101	2,4	23	107
44.070977.I430	485	1,4	95	200
45.070977.I600	343	0,5	93	57

concentration ppm(dry weight)

Samples	Zn(ppm)	Cd(ppm)	Pb(ppm)	Cu(ppm)
3I.080977.I0I0	II4	0,5	I5	90
4I.080977.0850	350	20 *	75	430 *
42.080977.0735	257	2,8	IO	I73
I3.03IO77.I240	34	37 *	I7	I55
23.03IO77.I4I5	340	II0 *	I23	343
33.03IO77.I745	422	I	I6	548 *
43.03IO77.I550	833	20 *	I25	837 *
II.04IO77.I045	494	63 *	IO6	407 *
II.05IO77.I550	400	0,7	77	364 *
2I.05IO77.I420	375	I6 *	69	II03 *
3I.05IO77.I250	2I40 *	2	338 *	2I82 *
4I 05IO77.II20	I296 *	84 *	I87	I724 *
I5.06IO77.I500	I364 *	I4 *	I54	I32
35.06IO77.I200	I505 *	5,5	99	I35
45.06IO77.II00	426	8,6	I47	I39
I4.07IO77.I300	27	I2 *	II4	I57
24.07IO77.II00	260	0,5	II6	99
34.07IO77.I000	2672 *	27,5 *	65I *	923 *
44.07IO77.0900	2I38 *	5	IO34 *	5648 *
I3.I2I277.I045	55	I,3	II	I8
23.I2I277.II45	I28	0,8	IO8	45

concentration ppm (dry weight)

Samples	Zn(ppm)	Cd(ppm)	Pb(ppm)	Cu(ppm)
33.I2I277.I350	31	0,7	12,5	34
43.I2I277.I300	68	0,4	128	62
15.I3I277.II25	136	3,2	84	51
25.I3I277.I225	114	3,2	50	-
35.I3I277.I430	89	0,05	73	56
45.I3I277.I340	42	0,3	12	25
11.I4I277.I030	15	0,5	8	11
21.I4I277.I205	56	2,2	11	45
31.I4I277.I445	18	0,4	3	13
41.I4I277.I345	13	0,4	15	49
12.I5I277.I200	49	0,3	58	46
22.I5I277.I240	36	1,2	7	72
32.I5I277.I450	5	0,3	12	25
42.I5I277.I400	20	3	10	28
14.I6I277.I030	256	0,5	-	-
24.I6I277.II20	91	0,1	47	33
44.I6I277.I235	32	0,2	5	8



important, however with considerable variations: 47-70% for Zn, 22-46% for Cd, 38-67% for Pb and 18-44% for Cu of the total dissolved metal. Higher levels for all of the metals are observed in the vicinity of the estuary(radial 5), on the radial I and in the coastal waters. The highest concentrations of Zn and Cd dissolved in seawater have been in samples collected during october 77, without significant changes for the other metals. This increase of Zn and Cd concentrations during this period are real and not due to contamination during the sampling. This situation is similar in july and december for Pb.

b)particulate metals.

Time variations in the particulate concentrations of Zn, Cd, Pb and Cu in the surface waters of the North Sea are presented in figure II and table VII.

As in the water samples, the highest values generally occurring in the vicinity of the estuary(points 15,35,45,34,44), Dunkerque(points 2I,3I, 4I,II) and the coast(points 13,23), increase to maximum mean values in autumn(october) and this is followed by a sharp drop to minimum concentrations between december and february.

These high levels are represented by obvious peaks in figure II and identified by an asterisk in table VII.

Unlike those in the water samples, the levels of all metals investigated in the particulate matter do tend to increase simultaneously in october.

CONCLUSION.

The increased levels observed for the metals(particulate and dissolved) during the cruise in october, coinciding with a period of high

productivity, suggests that plancton could be an additionnal factor leading to spacial complexity in the distribution of the metals.

However, in some instances higher values occured independently for the metal and not for the others. For exemple, on points II,4I,42,44, and I3 during july, Zn increases without significant changes in the other metals. The situation was similar in october on points II,2I, I3,23 and I4 for particulate Cd.

This fact could very well be related to differences in the lability of each element. However, we think that the factors governing the concentrations of soluble and particulate metals in the coastal waters in the North Sea are variable and more complex; metal levels in coastal surface waters appear to us to be more dependent on industrial dumping(points 34 and 23) and domestic waste than on biological factors.

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